DOI: https://doi.org/10.34287/MMT.1(44).2020.3

V. A. Kashirin, O. V. Khorolets, S. I. Andreev, A. A. Mikheev

 $State\ Institution\ «Zaporizhia\ Medical\ Academy\ of\ post-graduate\ education\ Ministry\ of\ Health\ of\ Ukraine \\ Zaporizhzhia\ Ukraine$

В. О. Каширін, О. В. Хоролець, С. І. Андрєєв, О. О. Міхєєв

Державний заклад «Запорізька медична академія післядипломної освіти Міністерства охорони здоров'я України» Запоріжжя, Україна

DYNAMICS OF THE VENOUS BLOOD ACID-BASE BALANCE AND RELATIONSHIP BLOOD PH VS TUMOR IN LARYNGEAL CANCER PATIENTS

Динаміка кислото-основного балансу венозної крові та взаємовідносини рН крові та пухлини у хворих на рак гортані

Abstract

The characteristic for most solid tumors cells is the intracellular alkalinization and acidification of the extracellular milieu and this pH gradient inversion (pHe < pHi) is associated with tumor proliferation, invasion, metastasis, aggressiveness, and treatment resistance. However is there tumor pH (pHi and/or pHe) changes affect on venous blood plasma pH?

Purpose of the study. The venous blood acid-base balance before and after the combined treatment, correlation of the venous blood pH indicators (pHb), relationship neoplasm and blood pH in patients with laryngeal cancer was study.

Material and methods. Studies were performed in patients with laryngeal cancer categories T2-3 NO MO before and after the combined treatment. The patients were divided into four groups: Group 1-25 patients before the start of treatment; Group 2-21 patients (from Group 1) after completion of the combined treatment; Group 3-14 patients from Group 2 with positive results of treatment and Group 4-7 patients from Group 2 with a negative result of treatment (recurrence and/or metastasis of the neoplasm). The control group consisted of 15 practically healthy people (Group C).

Examination of venous blood acid-base balance of patients, tumor pH and tumor cells pHi and pHe was carried.

Results and discussion. The increase in pCO_2 and HCO_3^- concentration will result in decrease in the pH, but if these indicators have a clear correlation in the control group, then in patients

Резюме

Для більшості солідних пухлин характерно підвищення рН цитоплазми клітин (рНі > 7,2) та підкислення позаклітинного середовища (рНе \approx 6,2-7,1). Зворотний рН градієнт ракових клітин (рНе < рНі) пов'язують з проліферацією, інвазією, метастазуванням та агресивністю пухлин але на цій особливості новоутворень розробляються і стратегії для підвищення ефективності їх лікування.

Мета дослідження. Вивчити кислотно-основний баланс венозної крові до і після комбінованого лікування, кореляцію показників рН венозної крові, взаємозв'язок новоутворення та рН крові у хворих на рак гортані.

Матеріали та методи. Дослідження виконані у хворих на рак гортані категорій Т2-3 NO MO до і після комбінованого лікування (період спостереження 2012–2018, період особистого спостереження не менше трьох років). Вік пацієнтів коливався від 34 до 65 років. Морфологічна характеристика новоутворень — зроговіла плоскоклітинна кариинома.

Хворі були розділені на чотири групи: група 1-25 пацієнтів до початку лікування; група 2-21 хворий (з групи 1) після завершення комбінованого лікування; група 3-14 пацієнтів з групи 2 з позитивними результатами лікування; група 4-7 хворих з групи 2 з негативним результатом лікування (рецидив та/або метастазування новоутворення).

Контрольну групу (група С) склали 15 практично здорових чоловіків того ж віку.

Проведено дослідження кислотно-основно-

groups there was a correlation for pHb & pCO $_2$ and pO $_2$ only. Besides, we marked increase in pCO $_2$, HCO_3^- , K^+ , while pO $_2$ decreased in pHb after the combined treatment.

It is necessary to point out the differences between some benchmarks and indicators of acid-base balance in the plasma of venous blood in primary patients and patients with recurrent laryngeal cancer. So, if pHb, pO_s, and Cl⁻ patients have statistically significant differences from control data, then differences with control pCO2 values are characteristic only for patients of Groups 1 and 3. On the contrary, differences in the HCO₃⁻ indices are characteristic only for patients of Group 4. There are statistically significant differences from the control indicators K^+ , Na^+ , Ca^{2+} , Glu, Lac, mOsmin patients of the first group and Cl- and Lac of patients in the third group. Among the indicators in the third and fourth groups of patients, statistically significant differences were noted in the values of pHb, HCO_3^- and Glu.

In patients of groups 1 and 4, the determination of pHt and the calculation of pHi, pHe revealed decrease in pHt and pHe with increasing pHi in patients with recurrence of the neoplasm.

The final stage of the study was to determine the relationship (and not correlation) of blood pH and laryngeal tumors and the relationship was noted in the «pHb-tumor» system in primary patients, but in patients in 3 and 4 Groups, that «pHb-tumor» connection is rather contradictory.

Conclusion. Acid-base balance indicators obviously cannot be considered as unconditional markers of carcinogenesis, but their monitoring and, in particular, venous blood pH, of patients after special treatment, can help determine the risk group of patients who may develop of a malignant neoplasm recurrence.

Keywords: acid-base balance, laryngeal cancer, relapse, prognosis.

го балансу венозної крові хворих. Визначений рН пухлин (pHt) з подальшим розрахунком рН цитоплазми клітин та рН позаклітинного середовища.

Результати та обговорення. Після завершення комбінованого лікування відзначено зниження рН плазми крові (рНb) та р O_2 при збільшені значень показників р CO_2 , HCO_3^- і K^+ . При порівнянні даних першої та третьої групи відмінності були відзначені лише в показниках Ca^{2+} але порівняння даних першої та четвертої групи показало значущі відмінності показників рHb, pO_2 , HCO_3^- і mOsm.

Збільшення концентрації pCO_2 і HCO_3^- призводе до зниження pH, але якщо ці показники мали чітку кореляцію в контрольній групі, то в першій, другій та третій групах кореляція відмічена лише для pHb/pCO_2 і pHb/pO_2 . У четвертій групі суттєвих кореляцій цих показників не відмічено.

Разом з тим, у пацієнтів 1 і 4 групи визначення рНt і розрахунок рНі, рНе виявили зниження рНt і рНе зі збільшенням рНі у пацієнтів з рецидивом новоутворення. Можливо це пояснює той факт, що кисле середовище мікрооточення сприяє прогресуванню пухлини — ацидоз, токсичний для нормальних клітин, сприяє деградації позаклітинного матриксу, але самі аномальні клітини стають менш вразливими.

Слід вказати на відмінності деяких контрольних показників і показників кислотно-лужного балансу в плазмі крові у первинних хворих і хворих з рецидивом раку гортані. Так, якщо показники pHb, pO_{\circ} и Cl^- хворих мають статистично значущі відмінності від контрольних даних, то відмінність з контрольними значеннями рСО характерна лише для хворих першої і третьої групи. Навпаки, відмінності показників НСО₀ - характерні лише для хворих четвертої групи. Мають статистично значущі відмінності $вi\partial$ контрольних показники K^+ , Na^+ , Ca^{2+} , Glu, Lac, mOsm хворих першої групи і Cl-u Lac хворих третьої групи. Серед показників в третій і четвертій групах хворих статистично значущі відмінності відзначені в значеннях показників р*Hb*, HCO₂- и Glu. При цьому, визначено наявність причинно-наслідкового зв'язку в системі «nyxлина-pHb» y первинних хворих, але y хворих з рецидивом захворювання залежність «пухлина-pHb» має досить суперечливий характер.

Висновок. Показники кислотно-основного балансу не можна розглядати як безумовні маркери канцерогенезу але їх моніторинг і зокрема рН венозної крові може допомогти визначити групу ризику серед пацієнтів, у яких може розвинутися рецидив злоякісного новоутворення.

Ключові слова: кислотно-основний баланс, рак гортані, рецидив, прогнозування.

INTRODUCTION

Cells population growth is determined by the genetic program of metabolism, wherein cells, for the order to function optimally, must constantly maintain an intracellular pH within a narrow range (pHi = 7,1-7,2) through the activity of transporters located at the plasma membrane. These transporters can be modulated by endogenous or exogenous molecules and pHi changes have been implicated in both cell proliferation and cell death. Notably, if the intracellular alkalinization is a common feature of proliferative processes, then the intracellular acidification has the cytotoxic effect through of apoptosis induction [1-3]. But for most cells of solid tumors characterized by the presence of intracellular alkalization of the cytoplasm (pHi > 7,2) and acidification of the extracellular milieu (pHe = 6,5-7,1) – a reverse pH gradient in cancer cells, unlike similar indicators of normal cells: pHi ≈ 7.2 , pHe ≈ 7.4 [4-8]. The acidification environmental tumors occurs already on early in cancers (cancer in situ), during the avascular phase, and pH gradient inversion (pHe < pHi) is associated with tumor proliferation, invasion, metastasis, aggressiveness, and treatment resistance [8-10].

Currently, it is believed that pH inversion, like hypoxia, is a common symptom of cancer and on this peculiarity of tumors various strategies are being built to increase the effectiveness of their treatment [11-18], but individual tumors may be had both positive and negative (reverse) pH gradients [19]. But still, the relationship between extracellular and intracellular pH are dependent upon the pH range. Intracellular pH was relatively resistant to a change in extracellular pH over the pHe range of 6,8 to 7,8 (i.e., delta pHi congruent to delta pHe \times 0,33) [20]. However is there tumor pH (pHi and/or pHe) changes affect on venous blood plasma pH?

$$\begin{aligned} pHi &= pHconst + \left[2 \times (pHconst - pHt)\right] \times 0,\!33 \\ pHe &= pHt - (pHi - pHconst) \end{aligned}$$

Comparisons of the studied parameters were carried out using the Wilcoxon Matched Pairs Test, Wald-Wolfowitz Runs Test and Spearman Rank Order Correlations, at the critical significance level of 0,05. The analyzed data are presented as «median and interquartile interval»: Me (RQ = UQ-LQ). The causal relationship between the indicators was evaluated using multiple logistic regression analysis. Statistical processing of the received data was made using computer programs of the STATISTICA package (StatSoft Statistica v.7.0).

RESULTS AND DISCUSSION

PURPOSE OF THE STUDY

The venous blood acid-base balance before and after the combined treatment, correlation of the venous blood pH indicators, relationship neoplasm and blood pH in patients with laryngeal cancer was study.

MATERIAL AND METHODS

Studies were performed in patients with laryngeal cancer categories T2-3 N0 M0 before and after the combined treatment (observation period 2012–2018, personal observation period after the combined treatment of at least three years). The age of the patients ranged from 34 to 65 years. The neoplasms morphological characteristic is keratinizing squamous cell carcinoma.

The patients were divided into four groups:

- Group 1 25 patients before the start of treatment.
- Group 2 21 patients (from Group 1) after completion of the combined treatment.
- Group 3 14 patients from Group 2 with positive results of treatment.
- Group 4 7 patients from Group 2 with a negative result of treatment (recurrence and/or metastasis of the neoplasm).

The control group consisted of 15 practically healthy people (Group C) of the same age.

Laboratory researches:

- 1. Examination of venous blood acid-base balance of patients with laryngeal cancer was carried – pH and pCO₂, pO₂ (mm/Hg), HCO₂-, K⁺, Na⁺, Ca², CL⁻, Glu, Lac, mOsm (mmol/l).
- 2. To the small ($\approx 2-2.5 \text{ mm}^3$) crushed tumor fragment was added 0.5 ml of $H_{\circ}O$ (pHconst = 7.328), the pH of the suspension was measured, which was regarded as the pH of the tumor.
- 3. Calculation of the pHi and pHe indicators was carried out according to the formula:

of patients, before the beginning and after treatment in table 1 are present.

The change in plasma pH in patients after treatment $(7,35 \rightarrow 7,31; p = 0,071)$ is not statistically significant, but it is important from a clinical point of view. But it should be noted, that all changes pH occur through changes in three independent variables - carbon dioxide, relative electrolyte concentrations, and total weak acid concentrations and some of these characteristics change and have statistically significant differences in groups 1 and 2 (increase in pCO₂ $46,08 \rightarrow 48,8$; HCO₃ $^ 24,9 \rightarrow 26,0$ and $K^{+}3,95 \rightarrow 4,5$, while pO₂ decreased 37,98 \rightarrow 23,63) [21].

However, the patients in Group 2 had both The data of pH venous blood plasma (pHb) positive (14 patients, who third Group constituted) and negative (7 patients) treatment results (Group 4). Moreover, when comparing the data of the first and third groups, statistically significant spills are noted in Ca^{2+} indicators only (1,14 \rightarrow 1,25), while comparing the data of the first and fourth groups,

significant differences were observed in indicators of pHb (7,35 \rightarrow 7,22) pO $_2$ (37,98 \rightarrow 20,0), HCO $_3$ (24,9 \rightarrow 22,2) \upmu mOsm (284,1 \rightarrow 288,4).

It is known, that increase in pCO_2 and HCO_3 concentration will result in decrease in the pH [21]:

$$pH = pK \times log [HCO_{3}^{-} / (0.03 \times pCO_{2})]$$

But if these indicators have a clear correlation in the control group, then in 1, 2, 3 groups of patient there was a correlation for pHb & pCO₂ and

pHb & pO₂ and no correlation for pHb & HCO₃⁻, and in the fourth group the insignificant correlation was noted for pHb & pCO₂ and pHb & pO₂ (table 2).

 ${\it Table~1}$ The acid-base balance indicators in the venous blood plasma of patients

Test	Group (Me; $RQ = UQ-LQ$)								
Test	Group 1 ($n = 25$)	Group 2 (n = 21)	Group 3 (n = 14)	Group 4 (n = 7)	p-level				
pHb	$7,35 \\ 7,42 - 7,31 = 0,11$	$7,31 \\ 7,34 - 7,31 = 0,03$	$7,32 \\ 7,38 - 7,31 = 0,07$	$7,29 \\ 7,31 - 7,26 = 0,05$	*0,071 +0,509 *0,091				
$\frac{\mathrm{pCO}_2}{\mathrm{mm/Hg}}$	$46,08 \\ 47,1-42,2=4,9$	$48,8 \\ 53,8-48,7=5,7$	$48,8 \\ 49,2-47,9=1,3$	56,1 62,9 - 48,7 = 14,2	*0,027 +0,331 *0,018				
$\begin{bmatrix} \mathrm{pO}_2\\ \mathrm{mm/Hg} \end{bmatrix}$	37,98 49,95 - 35,0 = 14,95	$23,63 \\ 29,4-19,0=10,4$	$29,0 \\ 32,0-19,0=13,0$	$19,0 \\ 20,0-14,1=5,9$	*0,002 *0,096 *0,063				
HCO-3 mmol/l	$24,9 \\ 25,19 - 24,65 = 0,54$	$26,0 \\ 28,7-25,1=3,6$	$25,02 \\ 26,0-24,8=1,2$	$25,8 \\ 29,0-20,7=8,3$	*0,017 +0,272 *0,799				
K ⁺ mmol/l	$3,95 \\ 4,13-3,9=0,23$	$4,5 \\ 5,1-4,1=4,0$	$4,35 \\ 5,0-4,0=1,0$	$4,9 \\ 5,2-4,5=0,7$	*0,009 +0,101 *0,063				
Na ⁺ mmol/l	$140,0 \\ 140,4-139,5=0,86$	$140,0 \\ 142,0-139,0=3,0$	139,5 142,0 - 139,0 = 3,0	$139,4 \\ 144,0-139,0=5,0$	*0,972 +0,949 *0,554				
Ca ²⁺ mmol/l	$1,14 \\ 1,2-1,07 = 0,13$	$1,19 \\ 1,26-1,17=0,09$	$1,25 \\ 1,27-1,18 = 0,09$	$1,22 \\ 1,27-1,0=0,27$	*0,106 +0,041 *0,866				
Cl ⁻ mmol/l	$107,0 \\ 109,0-107,0=2,0$	$108,0 \\ 109,0-107,7=1,3$	$109,0 \\ 109,0 - 104,0 = 5,0$	$108,0 \\ 111,0-108,0=3,0$	*0,834 +0,529 *0,753				
Glu mmol/l	$5,5 \\ 5,7-4,9=0,8$	$5,5 \\ 5,9-5,2=0,7$	$5,15 \\ 5,9-4,7=1,2$	$5,5 \\ 5,9-5,2=0,7$	*0,768 +0,754 *0,463				
Lac mmol/l	$1,65 \\ 1,81-1,5=0,31$	$ \begin{array}{c} 1,8 \\ 2,2-1,6=0,6 \end{array} $	$ \begin{array}{c} 1,75 \\ 2,2-1,5=0,7 \end{array} $	$2,0 \\ 2,3-1,5=0,8$	*0,112 +0,379 *0,643				
mOsm mmol/l	$284,1 \\ 284,5 - 283,8 = 0,7$	$285,6 \\ 289,1-283,0=6,1$	$284,05 \\ 289,1-282,0=7,1$	$288,0 \\ 291,5 - 284,5 = 7,0$	*0,395 +0,638 *0,176				

Note: *p-level for G1 Vs G2 groups; *p-level for G1 Vs G3 groups; *p-level for G1 Vs G4 groups

 $Table\ 2$ Correlations between plasma pH and pCO $_2$, pO $_2$ and HCO $_3$ in control and in groups of patients

				- Z' -	Z	3						
Spearman	Groups											
Rank Order Correlations	GC		(G1		G2		G3		G4		
	R	р	R	р	R	р	R	р	R	р		
$\mathrm{pHb} \;\&\; \mathrm{pCO}_{_{2}}$	0,53	0,042	-0,75	0,0001	-0,61	0,003	-0,74	0,002	-0,39	0,378		
$\mathrm{pHb} \;\&\; \mathrm{HCO}_{_{3}}$	0,56	0,029	-0,14	0,499	0,34	0,003	-0,07	0,819	-0,18	0,699		
pHb & pO ₂	0,72	0,0003	0,65	0,001	-0,61	0,003	0,91	0,000	0,38	0,398		

Changes in blood pH induce powerful regulatory effects at the level of the cell, organ, and organism, but how tumor pH (pH of cytoplasm + interstitial fluid) does it affect on blood pH and whether and to what extent it is possible to consider its dynamics as a predictor of the course of the disease or as performance evaluation of cancer patients treatment. However, we should highlight a few problems, associated with both the tumor and its microenvironment [7, 19–27].

Firstly, the resting pHi of a cell can be defined as the steady-state point at which net metabolic acid production is balanced by net membrane $\rm H^+/H^+$ -equivalent transport. But these fluxes to show considerable regional variation in solid tumors, resulting in the potential for large pHi gradients alongside pHe non-uniformity.

Second, tumor histology and tumor volume is the most important factors determining the range of pHe's. A combination of poor vasculature perfusion, regional hypoxia and increased flux of carbons through fermentative glycolysis leads to extracellular acidosis in solid tumors, with extracellular pH values as low as 6,5, but overall, actual pH in squamous cell

carcinomas is 7.20 ± 0.07 (pHt in range 6.2-7.6) with the pHi and pHe values lying mostly in the range 7.1-7.65 and 6.2-6.9 respectively [28–32].

Thirdly, tissue pH is difficult to investigate by measuring pH in cells suspensions or monolayers prepared from cultured cells. Besides no equations which would allow accurate calculation of indicators pHi or pHe based on the pHt indicator. Rather, these calculations will confirm the trend – pHi and pHe will change in opposite directions (extracellular acidification and intracellular alkalinization).

In patients of groups 1 and 4, the determination of pHt and the calculation of pHi, pHe revealed decrease in pHt and pHe with increasing pHi in patients with recurrence and/or metastasis of the neoplasm (table 3). Obviously this can be explained by surviving in treatment tumor cells begin to actively proliferate and the acidic environment of the microenvironment contributes to tumor progression, stimulating invasion and metastasis, acidosis can be toxic to normal cells and mediate degradation and remodeling of the extracellular matrix, can enhance angiogenesis due to release of vascular endothelial growth factor, but themselves abnormal cells become less vulnerable [31, 33–37].

 $Table\ 3$ Tumor pH indicators in patients of the 1 and 4 groups

	Group (Me; R			
Test		p-level		
	Group 1 (n = 25)	Group 4 (n =7)		
pHt	$7,05 \\ 7,09-7,02 = 0,07$	$6,98 \\ 7,035-6,915 = 0,12$	0,002	
рНе	$6,87 \\ 6,852-6,712=0,14$	$6,75 \\ 6,742-6,502=0,24$	0,002	
pHi	$7,51 \\ 7,636-7,596 = 0,04$	$7,56 \\ 7,741-7,621 = 0,12$	0,057	

It is necessary to point out the differences between some benchmarks and indicators of acid-base balance in the plasma of venous blood in primary patients and patients with recurrent laryngeal cancer. So, if pHb, pO₂, and Cl⁻ patients have statistically significant differences from control data, then differences with control pCO₂ values are characteristic only for patients of Groups 1 and 3. On the contrary, differences in the HCO₂ indices are characteristic only for patients of Group 4. There are statistically significant differences from the control indicators K⁺, Na⁺, Ca²⁺, Glu, Lac, mOsm in patients of the first group and Cl- and Lac of patients in the third group. Among the indicators in the third and fourth groups of patients, statistically significant differences were noted in the values of pHb, HCO_3^- and Glu (table 4).

Thus, the presented data indicate the presence of significant discrepancies in the control values of a number of indicators of acid-base balance and indicators of acid-base balance of patients who successfully completed treatment, from the corresponding indicators of primary patients and patients with recurrent neoplasm.

However, how pHb is coupled to cancer cell growth? The final stage of the research was the determination of the relationship (not correlation) of blood pH and laryngeal tumors and it is necessary to recall that the odds ratio (OR) from 0 to 1 indicates a low probability of the event being investigated, the OR of 1 means that the likelihood of an event is the same in both groups. The greater the odds ratio unit, the more likely it is to expect an even to develop (table 5). The analysis is carried out in groups G1 & GC, G4 & GC and G3 & G4.

From the data presented in the table indicate the presence of a causal relationship in the «pHb – tumor» system in primary patients, but in patients in 3 and 4 Groups, the «pHb – tumor» connection is rather contradictory. Obviously, this can be explained both by the presence of a progressive neoplasm and by the aggressive nature of the treatment methods carried.

 $Table\ 4$ The acid-base balance indicators in the venous blood plasma in the control group and in the first, third and fourth group patients

TTd	Group (Me; RQ = UQ-LQ)								
Test	Group C (n = 15)	Group 1 (n = 25)	Group 3 (n = 14)	Group 4 (n = 7)	p-level				
pHb	$7,39 \\ 7,41-7,36 = 0,05$	7,35 7,42 - 7,31 =0,11	$7,32 \\ 7,38 - 7,31 = 0,07$	$7,22 \\ 7,31-7,11=0,2$	*0,002 *0,038 +0,0001 °0,003				
pCO_2	46.7 $48.2 - 45.8 = 2.4$	$46,08 \\ 47,1-42,2=4,9$	48,8 $49,2-47,9=1,3$	48,8 58,8 - 48,3 = 10,5	*0,000 ×0,038 +0,299 °0,673				
pO_2	$44,2 \\ 45,5 - 42,5 = 3,0$	37,98 $49,95 - 35,0 = 14,95$	$ 29,0 \\ 32,0-19,0=13,0 $	$20,0 \\ 22,5-19,0=3,5$	*0,000 *0,0007 +0,0001 °0,353				
HCO ₃	$24,1 \\ 25,0-24,1=0,9$	24,9 $25,19 - 24,65 = 0,54$	$25,02 \\ 26,0-24,8=1,2$	$22,2 \\ 23,8-20,7=3,1$	*0,266 *0,574 +0,040 °0,052				
K ⁺	5,0-4,4=0,6	3,95 $4,13-3,9=0,23$	$4,35 \\ 5,0-4,0=1,0$	$ 4,9 \\ 5,2-4,5=0,7 $	*0,005 *0,347 +0,982 *0,151				
Na ⁺	142,0 144,0 - 140,0 = 4,0	140,0 $140,4-139,5=0,86$	$139,5 \\ 142,0-139,0=3,0$	139,2 144,0 - 139,0 = 5,0	*0,032 *0,347 +0,628 °0,933				
Ca^{2+}	$1,28 \\ 1,31-1,24=0,07$	1,14 $1,2-1,07 = 0,13$	1,25 $1,27-1,18=0,09$	$1,25 \\ 1,31-1,02=0,29$	*0,032 *0,187 +0,628 *0,151				
Cl-	$ \begin{array}{c c} 105,0 \\ 107,0-102,0=5,0 \end{array} $	107,0 $109,0-107,0=2,0$	$ \begin{array}{c c} 109,0 \\ 109,0 - 104,0 = 5,0 \end{array} $	$109,0 \\ 111,0-108,0=3,0$	*0,013 *0,038 +0,040 °0,353				
Glu	5,1-4,4=0,7	$5,5 \\ 5,7-4,9=0,8$	5,15 $5,9-4,7=1,2$	$5,9 \\ 5,9-5,2=0,7$	*0,072 *0,574 +0,040 °0,544				
Lac	$2,1 \\ 2,4-1,9=0,5$	$1,65 \\ 1,81-1,5=0,31$	$ \begin{array}{c} 1,75 \\ 2,2-1,5=0,7 \end{array} $	$ 2,0 \\ 3,2-1,5=1,7 $	*0,032 *0,038 +0,319 *0,933				
mOsm	$288,0 \\ 291,1-286,0=5,1$	$284,1 \\ 284,5 - 283,8 = 0,7$	$284,05 \\ 289,1 - 282,0 = 7,1$	288,4 290,0 - 284,5 = 5,5	*0,000 *0,187 *0.077 *0,673				

 $\textbf{Note: *p-level for GC Vs G3 groups; *p-level for GC Vs G3 groups; *p-level for GC Vs G4 groups; *p-level for G3 Vs G4 groups; *p-level for GC Vs G4 groups; *p-level for G5 Vs G$

Logistic regression												
Test	G1 & GC			G4 & GC			G3 & G4					
	pН	pCO ₂	\mathbf{pO}_2	HCO ₃ -	pН	pCO ₂	\mathbf{pO}_2	HCO ₃ -	pН	\mathbf{pCO}_2	\mathbf{pO}_2	HCO ₃
Estimate	11,91	0,026	0,025	0,909	0,94	0,67	2,96	1,03	6,5E+01	6E+01	0,17	1,5E+0
OR (unit ch)	141312	1,026	1,026	0,402	0,39	0,51	19,5	2,8			1,18	4,4E+0
OR (range)	12,95	1,619	4,109	0,013	0,65	0,01		366	4,5E+10	5E+10	218	4,5E+0,5

CONCLUSION

Acid-base balance indicators obviously cannot be considered as unconditional markers of carcinogenesis,

but their monitoring and, in particular, venous blood pH, of patients after special treatment, can help determine the risk group of patients who may develop of a malignant neoplasm recurrence.

REFERENCE

- 1. DiGiammarino EL, Lee AS, Cadwell C, Zhang W et al. A novel mechanism of tumorigenesis involving pH-dependent destabilization of a mutant p53 tetramer. Nat. Struct. Biol. 2002; 9 (1): 12-16. DOI: 10.1038/nsb730.
- 2. Capuano P, Capasso G. The importance of intracellular pH in the regulation of cell function. G Ital Nefrol. 2003; 20 (2): 139-150.
- 3. Legadic-Gossmann D, Huc L, Lecureul V. Alterations of intracellular pH homeostasis in apoptosis: origins and roles. Cell Death Differ. 2004; 11 (9): 953–961. DOI: 10.1038/sj.cdd.4401466
- 4. Helmlinger G, Yuan F, Dellian M, Jain, RK. Interstitial pH and pO2 gradients in solid tumors in vivo: high-resolution measurements reveal a lack of correlation. Nat. Med. 1997; 3 (2): 177–182.
- 5. Casey, JR, Grinstein, S, Orlowski J. Sensors and regulators of intracellular pH. Nat. Rev. Mol. Cell Biol. 2010; 11 (1): 50-61. DOI: 10.1038/nrm2820.
- 6. Webb BA., Chimenti M, Jacobson MP, Barber DL. Dysregulated pH: a perfect storm for cancer progression. Nat Rev Cancer. 2011; 11 (9): 671–677. DOI: 10.1038/nrc3110
- 7. Persi E, Duran-Frigola M, Damaghi M, Roush WR et al. Systems analysis of intracellular pH vulnerabilities for cancer therapy. Nat Commun. 2018; 9 (1): 2997. DOI: 10.1038/s41467-018-05261-x.
- 8. Barathova M, Takacova M, Holotnakova T, Gibadulinova A et al. Alternative splicing variant of the hypoxia marker carbonic anhydrase IX expressed independently of hypoxia and tumour phenotype. Br J Cancer. 2008; 98 (1): 129-136. DOI: 10.1038/sj.bjc.6604111.
- 9. Damaghi M, Wojtkowiak JW, Gillies RJ. pH sensing and regulation in cancer. Front Physiol. 2013; 4: 370. DOI: 10.3389/fphys.2013.00370.
- 10. Kulikov VA, Belyaeva LE. On bioenergetics of a tumoral cell. Vestnik vitebskogo gosudarstvennogo meditsinskogo universiteta. 2015; 14 (6): 5–14.
- 11. Ludwig MG, Vanek M, Guerini D, Gasser JA et al. Proton-sensing G-protein-coupled receptors. Nature. 2003; 425 (6953): 93–98. DOI: 10.1038/nature01905.
- 12. Bumke MA, Neri D, Elia G Modulation of gene expression by extracellular pH variations in human fibroblasts: a transcriptomic and proteomic

- study. Proteomics. 2003; 3 (5): 675–688. DOI: 10.1002/pmic.200300395.
- 13. Chiche J, Brahimi-Horn MC, Pouysségur J. Tumour hypoxia induces a metabolic shift causing acidosis: a common feature in cancer. J Cell Mol Med. 2010; 14 (4): 771-794. DOI: 10.1111/j.1582-4934.2009.00994.x.
- 14. McCarty MF, Whitaker J. Manipulating tumor acidification as a cancer treatment strategy. Altern Med Rev. 2010; 15 (3): 264–272.
- 15. Miranda-Gonçalves V, Reis RM, Baltazar F. Lactate Transporters and pH Regulation: Potential Therapeutic Targets in Glioblastomas. Curr Cancer Drug Targets. 2016; 16 (5): 388-399.
- 16. Corbet C, Feron O. Tumour acidosis: from the passenger to the driver's seat. Nat. Rev. Cancer. 2017; 17 (10): 577-593. DOI: 10.1038/nrc.2017.77.
- 17. Granja S, Tavares-Valente D, Queiros O, Baltazar F. Value of pH regulators in the diagnosis, prognosis and treatment of cancer. Semin Cancer Biol. 2017; 43: 17-34. DOI: 10.1016/j.semcancer.2016.12.003.
- 18. Parks SK, Pouyssegus J. Targeting pH regulating proteins for cancer therapy-Progress and limitations. Semin Cancer Biol. 2017; 43: 66–73. DOI: 10.1016/j.semcancer.2017.01.007.
- 19. Prescott DM, Charles HC, Poulson JM, Page RL et al. The relationship between intracellular and extracellular pH in spontaneous canine tumors. Clin Cancer Res. 2000; 6 (6): 2501-2505.
- 20. Fellenz MP, Gerweck LE. Influence of extracellular pH on intracellular pH and cell energy status: relationship to hyperthermic sensitivity. Radiat Res. 1988; 116 (2): 305–312.
- 21. Kellum JA. Determinants of blood pH in health and disease. Crit Care. 2000; 4 (1): 6-14. DOI: 10.1186/cc644.
- 22. Capuano P, Capasso G. The importance of intracellular pH in the regulation of cell function. G Ital Nefrol. 2003; 20 (2): 139-150.
- 23. Hanahan, D. & Weinberg, R. A. Hallmarks of cancer: the next generation. Cell. 2011; 144 (5), 646–674. DOI: 10.1016/j.cell.2011.02.013.
- 24. Criscitiello C, Esposito A, Curigliano G. Tumor-stroma crosstalk: targeting stroma in breast cancer. Curr Opin Oncol. 2014; 26 (6): 551–555. DOI: 10.1097/CCO.0000000000000122.

- 25. Swietach P, Vaughan-Jones RD, Harris AL, Hulikova A. The chemistry, physiology and pathology of pH in cancer. Philos Trans R Soc Lond B Biol Sci. 2014; 396 (1638): 20130099. DOI: 10.1098/rstb.2013.0099.
- 26. Huber V, Camisaschi C, berzi A, Ferro S et al. Cancer acidity: An ultimate frontier of tumor immuneescapeandanoveltargetofimmunomodulation. Seminars in Cancer Biology. 2017; 43: 74–89. DOI: 10.1016/j.semcancer.2017.03.001.
- 27. Puppulin L, Hosogi S, Sun H, Matsuo K et al. Bioconjugation strategy for cell surface labelling with gold nanostructures designed for highly localized pH measuremen. Nature Communications. 2018; 9, Article number: 5278. DOI: 10.1038/s41467-018-07726.
- 28. Engin K, Leeper DB, Cater JR et al. Extracellular pH distribution in human tumours. Int J Hyperthermia. 1995; 11 (2): 211-216.
- 29. Swietach P, Vaughan-Jones RD, Harris AL. Regulation of tumor pH and the role of carbonic anhydrase 9. Cancer Metastasis Rev. 2007; 26 (2): 299-310. DOI: 10.1007/s10555-007-9064-0.
- 30. Swietach P, Hulikova A, Vaughan-Jones RD, Harris AL. New insights into the physiological role of carbonic anhydrase IX in tumour pH regulation. Oncogene. 2010; 29 (50): 6509–6521. DOI: 10.1038/ONC.2010.455.
- 31. Damaghi M, Tafreshi NK, Lloyd MC, Sprung R et al. Chronic acidosis in the tumour

- microenvironment selects for overexpression of LAMP2 in the plasma membrane. Nat Commun. 2015; 6: 8752. DOI: 10.1038/ncomms9752
- 32. Morishima H, Jumpei Washio J, Kitamura J, Shinohara Y et al. Real-time monitoring system for evaluating the acid-producing activity of oral squamous cell carcinoma cells at different environmental pH. Scientific Reports 7, Article number: 10092 (2017). DOI: 10.1038/s41598-017-10893-y.
- 33. Harris AL. Hypoxia a key regulatory factor in tumor growth. Nature Rev Cancer. 2002; 2 (1): 38–47. DOI: 10.1038/nrc704
- 34. Cheng C, Sharp PA. Regulation of CD44 alternative splicing by SRm160 and its potential role in tumor cell invasion. Mol Cell Biol. 2006; 26 (1): 362–370. DOI:10.1128/MCB.26.1.362-370.2006.
- 35. Prochazka L, Tesarik R, Turanek J. Regulation of alternative splicing of CD44 in cancer. Cell Signal. 2014; 26 (10): 2234-2239. DOI: 10.1016/j.cellsig.2014.07.011.
- 36. Cali B, Molon B, Viola A. Tuning cancer fate: the unremitting role of host immunity. Open Biol. 2017; 7 (4): pii: 170006. DOI: 10.1098/rsob.170006.
- 37. Marquardt S, Solanki M, Spitschak A, Vera J, PützerBM. Emerging functional markers for cancer stem cell-based therapies: Understanding signaling networks for targeting metastasis. Semin Cancer Biol. 2018; 53: 90-109. DOI: 10.1016/j.semcancer.2018.06.006.

Стаття надійшла до редакції 16.09.2019